State-independent Block of BK Channels by an Intracellular Quaternary Ammonium

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Intracellular blockade by quaternary ammonium (QA) molecules of many potassium channels is state dependent, where the requirement for channel opening is evidenced by a time-dependent component of block in the macroscopic record. Whether this is the case for Ca\(^{2+}\)- and voltage-activated potassium (BK) channels, however, remains unclear. Previous work (Li, W., and R.W. Aldrich. 2004. J. Gen. Physiol. 124:43–57) tentatively proposed a state-dependent, trapping model, but left open the possibility of state-independent block. Here, we found BK channel blockade by a novel QA derivative, bbTBA, was time dependent, raising the possibility of state-dependent, open channel block. Alternatively, the observed voltage dependence of block could be sufficient to explain time-dependent block. We have used steady-state and kinetic measurements of bbTBA blockade in order to discriminate between these two possibilities. bbTBA did not significantly slow deactivation kinetics at potentials between −200 and −100 mV, suggesting that channels can close unhindered by bound bbTBA. We further find no evidence that bbTBA is trapped inside BK channels after closing. Measurements of steady state fractional block at +40 mV revealed a 1.3-fold change in apparent affinity for a 33-fold change in P\(_{\text{off}}\) in striking contrast to the 31-fold change predicted by state-dependent block. Finally, the appearance of a third kinetic component of bbTBA blockade at high concentrations is incompatible with state-dependent block. Our results suggest that access of intracellular bbTBA to the BK channel cavity is not strictly gated by channel opening and closing, and imply that the permeation gate for BK channels may not be intracellular.

INTRODUCTION

The mechanism of ion channel opening and closing is widely thought of in terms of a gate that permits or restricts access to the inner pore via a reversible conformational change. Much attention has been focused on locating the gate of K\(^{+}\) channels, as a first step toward understanding the conformational changes that underlie the gating process. Clay Armstrong’s innovative use of intracellular organic blockers that could block K\(^{+}\) channels only when they were open (Armstrong, 1966, 1969, 1971) produced the first strong evidence for an intracellular activation gate. The classic features of state-dependent (open channel) block defined in this work (Armstrong, 1971) include (a) time-dependent block of macroscopic current, where steady-state block reaches equilibrium only after open probability is sufficiently high, (b) a “foot in the door” effect of blocker on channel closing, where internally bound quaternary ammonium (QA) ions of a sufficient size were found to delay closure of the activation gate, and (c) “trapping” of a blocker molecule inside the channel pore after closure of the activation gate. These observations strongly suggested that intracellular blockers required channel opening in order to access (and exit) the pore, and implied that the gating structure regulating access of the blocker to the pore must be located on the intracellular face of the channel. Numerous functional studies of K\(^{+}\) channels (Armstrong and Hille, 1972; MacKinnon and Miller, 1989; MacKinnon and Yellen, 1990; MacKinnon et al., 1990; Yellen et al., 1991; Heginbotham et al., 1992; Liu et al., 1997; Shin et al., 2001) and structural studies of bacterial (Doyle et al., 1998; Jiang et al., 2002a; Jiang et al., 2003; Kuo et al., 2003) and mammalian (Long et al., 2005) K\(^{+}\) channels have confirmed Armstrong’s original hypothesis.

The structure of the mammalian voltage-gated K\(^{+}\) channel Kv 1.2 (Long et al., 2005) exemplifies the detailed organization of an archetypal K\(^{+}\) channel pore. The S5 and S6 transmembrane domains from each of four subunits define the central pore, a large water-filled cavity constrained at the extracellular end by the narrow selectivity filter, and at the intracellular end by the “bundle crossing,” where the S6 helices come into close apposition. For the majority of K\(^{+}\) channels, permeation of K\(^{+}\) ions appears to be governed by constriction and enlargement of the S6 aperture. On the other hand, it is clear that some related ion channels employ a different gating mechanism. The case for cyclic

Abbreviations used in this paper: bbTBA, N-(4-[benzoyl]benzyl)-N,N,N,N-tetra-n-butylammonium bromide; BK, large-conductance Ca\(^{2+}\)-and voltage-activated potassium; C10, decapeptylaminium; CNG, cyclic nucleotide-gated; P\(_{\text{off}}\), open probability; QA, quaternary ammonium; TBA, tetrabutylammonium.
nucleotide-gated (CNG) channels is particularly compelling, where Flynn and Zagotta (2001) found that although access of internally applied MTSET to pore residues depends upon the state of the channel, those residues are equally accessible to internal Ag⁺ (roughly equivalent to a K⁺ ion in size) in the open or closed state. The authors propose that although the larger MTS reagent is gated by conformational changes of the S6 bundle crossing, permeant ions may be gated at the selectivity filter for CNG channels. Similarly, small-conductance Ca²⁺-activated K⁺ (SK) channels (Bruening-Wright et al., 2002), ATP-sensitive inward rectifier K⁺ (KᵢATP) channels (Proks et al., 2003), and KcsA K⁺ channels (Cordero-Morales et al., 2006) may also be gated at the selectivity filter.

For BK Ca²⁺- and voltage-activated K⁺ channels, the location of the permeation gate remains an open question. Although the structure of a bacterial Ca²⁺-gated K⁺ channel (MthK) is known (Jiang et al., 2002a), and functional evidence from BK channels is consistent with motion of the intracellular S6 domains during gating (Niu et al., 2004), the mechanism of BK channel gating is far from fully understood. In some ways, intracellular block of BK channels appears similar to classical open channel block. Most notably, block of BK channels by intracellular Shaker ball peptide is state dependent; the peptide blocks only open channels and must exit before channels can close (unpublished data). Several other observations could also be consistent with an intracellular gate for BK channels: mutations in S6 disrupt activation gating (Piskorowski, 2005), BK channels can close when the selectivity filter is occupied by either Na⁺ or Ba²⁺ (Miller et al., 1987; Neyton and Pelleschi, 1991), and high concentrations of external K⁺ have no obvious slowing effect on deactivation of BK channels (Neyton and Pelleschi, 1991; Demo and Yellen, 1992).

In spite of these observations, however, there is equally compelling evidence suggesting that the BK channel gate may lie elsewhere. First, QA molecules, which have been shown functionally to promote C-type inactivation (a form of selectivity filter gating in Kv channels; Choi et al., 1991) and are proposed to induce constriction of the intracellular Shaker ball peptide (provided by the β subunit) does not require channels to visit the fully conducting open state (Benzenziger et al., 2006). Third, no evidence has been obtained thus far for trapping of large intracellular blocker molecules like leupeptin, curare, QX-314, and chlorisondamine (Solaro et al., 1997). Fourth, external Rb⁺ has been shown to destabilize the closed state, presumably due to occupancy of the selectivity filter (slowing deactivation and speeding channel opening; Demo and Yellen, 1992). Finally, a novel conotoxin, K-BtX, appears to stabilize the open state of BK channels from the external side (Fan et al., 2003). It is thus not unreasonable to consider the possibility that BK channels are gated at the selectivity filter, particularly in light of the fact that three other channels activated by intracellular ligands, the CNG, KᵢATP and SK channels, appear to use selectivity filter gates (Flynn and Zagotta, 2001; Bruening-Wright et al., 2002; Proks et al., 2003).

The classic QA blocker experiments that have worked so well for other K⁺ channels have been less informative for BK channels because of the relatively large size of the BK channel intracellular cavity. In order for a time-dependent component of block to be visible, the kinetics of block must be slow with respect to the intrinsic rate of channel opening, and the kinetics of block by even the largest molecule tested thus far, decyltriethylammonium (C10), have been too fast to resolve (Li and Aldrich, 2004). We sought to find a QA blocker with sufficiently slow kinetics to allow resolution of time-dependent block and to discern whether intracellular block of BK channels requires channel opening. Here we provide evidence that the QA ion bbTBA (N-(4-[[benzoyl]benzyl]-N,N,N-tributylammonium bromide) can block both closed and open channels. We conclude that access of bbTBA to the BK channel pore does not appear to be strictly regulated by an intracellular gating structure, and speculate that permeant K⁺ ions may be gated in the selectivity filter itself.

**MATERIALS AND METHODS**

**Channel Expression**

All experiments were conducted using the mSlo gene encoding the mouse BK channel (Butler et al., 1993). Freshly isolated Xenopus oocytes were injected with ~5 ng or ~0.05 ng cRNA for macroscopic or microscopic experiments, respectively. Patch clamp recordings were made 1–4 d after injection.

**Electrophysiology**

All recordings were made from oocyte macropatches using the inside-out patch configuration (Hamill et al., 1981). Pipettes were pulled from borosilicate glass (VWR Scientific) to yield resistances of 1–3 MΩ, and coated with sticky wax (Kerr Corporation). All experiments were performed at 22°C. Data were acquired with an Axopatch 1B or Axopatch 200A amplifier (Axon Instruments, Inc.), low pass filtered at 10 kHz, and digitized using an ITC-16 converter (Instrutech). Sampling interval was set at 10 or 20 μs using Pulse software (HEKA Electronik) and a Macintosh G3 computer. Data were leak subtracted using a −P/4 protocol.
stepping from leak holding potential of \(-120\) mV. Macroscopic ionic currents were usually averaged from three or more consecutive sweeps. Idealization of microscopic recordings was accomplished using the TAC software suite (Bruxton Corporation); all further analysis of macroscopic and microscopic data was performed with Igor (WaveMetrics, Inc.).

### Analysis and Curve Fitting

Dose–response data were fitted with the Hill equation:

\[
1 - \frac{I_{\text{block}}}{I_{\text{control}}} = 1/(1 + (K_h/([\text{bbTBA}])))^n \tag{1}
\]

where \(I_{\text{block}}\) and \(I_{\text{control}}\) are current with and without bbTBA, respectively, \(K_h\) is apparent dissociation constant for bbTBA, and \(n = \) Hill coefficient. The voltage dependence of \(k_{\text{OFF}}\), \(k_{\text{ON}}\), or \(K_h\) was determined by fitting with the Woodhull equation (Woodhull, 1973):

\[
k(V) = k(0)e^{z\Delta V/RT} \tag{2}
\]

where \(V\) = test potential, \(k(0)\) = the rate at \(0\) mV, \(z\) = valence of bbTBA (+1), and \(\Delta V\) = effective electrical distance; together, \(z\) closely approximates the voltage dependence of blockade. Voltage dependence of block was also measured by fitting fractional unblock as a function of voltage with a Woodhull equation of the form

\[
I_{\text{block}}/I_{\text{control}} = 1/(1+([\text{bbTBA}]/K_h(0)e^{-z\Delta V/RT})) \tag{3}
\]

### Conductance–voltage relationships

Conductance–voltage relationships were calculated from isochronal tail current measurements at \(-80\) mV after variable test steps, and normalized to maximal tail current amplitude. This procedure avoids complications induced by fast Ca\(^{2+}\) block of steady-state currents, and thus accurately reflects relative macroscopic open probability (\(P_o\)) as a function of voltage (Cox, Cui and Aldrich, 1997). The data were fitted with a Boltzman function:

\[
G(V) = G_{\text{max}}/(1 + \exp(-zF(V - V_{1/2})/RT)) \tag{4}
\]

where \(G\) = conductance, \(V_{1/2}\) = potential for half-maximal activation, and \(z\) = apparent equivalent gating charge. Current traces depicted in the figures were zero subtracted for display purposes. In general, the SEM is given with averaged values, while the 95% confidence interval is provided for fitted parameters.

### Predictions of Percent Block

#### Competitive Block

The apparent \(K_{\text{app}}\) \(K_{\text{app}}\) for bbTBA or tetra-butylammonium (TBA) were estimated from the average fractional block (\(F\)) observed for each blocker alone using \(F = [\text{QA}]/[\text{QA} + K_{\text{app}}]\). This yielded \(K_{\text{app}}\), where \(K_{\text{app}}\) \(K_{\text{app}}\) was obtained from either kinetic measurements (Fig. 4 C) or steady state \(P_o\) (Fig. 4 D). Blocker rates and values were obtained from fitting plots of either the on or off rates from dwell time histograms vs. voltage (Fig. 5, A and B) with Eq. 2.

A final model was constructed that describes all measurements in this study; this model was used in modeling block (Figs. 1, 3, 5, 8, 10, and 11; see Materials and Methods).

### Table I

<table>
<thead>
<tr>
<th>Parameters of bbTBA Blockade</th>
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<tr>
<td>Macroscopic</td>
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<td>Frac. block (Fig. 1 C)</td>
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<td>Kinetics state (Fig. 5 C)</td>
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<td>Steady state (Fig. 5 C)</td>
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<tr>
<td>(K_h(0\text{ mV})) (\mu M)</td>
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<tr>
<td>9.5</td>
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<tr>
<td>8.3</td>
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<tr>
<td>8.1</td>
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<td>10.1</td>
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<tr>
<td>0.13</td>
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<tr>
<td>(k_{\text{OFF}}(0\text{ mV})) (s^{-1})</td>
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<tr>
<td>(k_{\text{ON}}(0\text{ mV})) (s^{-1})</td>
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<td>(562)</td>
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<tr>
<td>(z\beta_{\text{OFF}})</td>
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The dissociation constant \(K_h\) reflects the concentration of bbTBA required for half-maximal blockade. The charge carried by bbTBA (\(z = +1\)) multiplied by the total electrical distance \(\delta\) together give \(z\beta_{\text{OFF}}\), an estimate of the total voltage dependence of block. The second order association rate for block at \(0\) mV \(k_{\text{ON}}(0\text{ mV})\) and the charge associated with blocker binding \(z\beta_{\text{ON}}\), as well as the rate and voltage dependence of blocker dissociation \(k_{\text{OFF}}(0\text{ mV})\) and \(z\beta_{\text{OFF}}\), were calculated from microscopic kinetic measurements as described in Figs. 4 and 5. Macroscopic \(K_h\) and \(z\) were obtained from fitting \(K_h\) values (obtained from Hill fits) plotted as a function of voltage (Fig. 1 C) with Eq. 2, or fitting fractional unblock plotted as a function of voltage (Fig. 8 C) to Eq. 3. Macroscopic \(K_h\) and \(z\) were measured from fits to Eq. 2 of \(K_h\) vs. voltage relationships (Fig. 5 C), where \(K_h\) was obtained from either kinetic measurements (Fig. 4 C) or steady state \(P_o\) (Fig. 4 D). Blocker rates and associated \(z\) values were obtained from fitting plots of either the on or off rates from dwell time histograms vs. voltage (Fig. 5, A and B) with Eq. 2.

A final model was constructed that describes all measurements in this study; this model was used in modeling block (Figs. 1, 3, 5, 8, 10, and 11; see Materials and Methods).

#### Noncompetitive Block

Percent block was calculated by summing the fraction of current blocked by TBA, \(F_{\text{TBA}}\), and the fraction of current remaining in the presence of TBA that is blocked by bbTBA, \(F_{\text{bbTBA}}\).

\[
F_{\text{bbTBA}} = (1 - F_{\text{TBA}})F_{\text{bbTBA}} \tag{5}
\]

Final Model for bbTBA Blockade

A final model for state-independent blockade of BK channels by bbTBA was constructed (Scheme 1), which accurately reproduced steady-state and kinetic features of block (see Figs. 1, 3, 5, 8, 10, and 11).

![Scheme 1](image_attachment)

This model, based on the simulation protocol developed by Horrigan et al. (1999), consisted of four connected states, with parameters taken from our average macroscopic and microscopic data (see Table I). The final \(z\) value of 0.15 was calculated from the average values of \(z\) derived from six different measurements, and closely matches that observed for C10 and TBA (~0.16 and ~0.21; Li and Aldrich 2004). \(K_h(0\text{ mV})\) was allowed to vary somewhat (8–10 \(\mu M\)) in order to compensate for the observed shift in \(K_h\) over time and variability in individual patches. Because macroscopic measurements were made earlier in time, the average on and off rates are biased toward the lower \(K_h(0\text{ mV})\) observed at that time (~5 \(\mu M\); Table I).

Assuming that the decline in apparent affinity of bbTBA over time is the result of a degradative process, we expect that the increased \(K_h\) reflects a decrease in \(k_{\text{OFF}}\) with relatively little change in \(K_{\text{app}}\). Thus, in order for our final model to accommodate measurements made later in time, we constrained the model to the macroscopic steady-state parameters \(K_h(0\text{ mV}) = 8–10\) \(\mu M\) and \(z = 0.15\), and also required that \(k_{\text{OFF}}(0\text{ mV}) = 502 \text{ s}^{-1}\), while \(k_{\text{ON}}(0\text{ mV})\) was scaled to match the \(K_h(0\text{ mV}) = 0.625 \times 10^8\) \(\text{M}^{-1}\text{ s}^{-1}\). Microscopic measurements of the voltage dependence of \(K_{\text{app}}(0.05\text{ mV})\) and \(k_{\text{OFF}}(0.7\text{ mV})\) were assumed to be accurate, and \(k_{\text{OFF}}\)
was adjusted slightly (to 0.1) to yield \( z_{\text{model}} = 0.15 \). Note that without these adjustments, the kinetic features of block, including the triphasic time dependence observed at high concentrations, were still reproducible. Our final model makes two assumptions: (1) \( \text{bbTBA} \) binding does not alter the intrinsic rates of channel activation and deactivation (e.g., the same values for \( \alpha \) and \( \beta \) were used for the closed/open transition for both unblocked and blocked channels), and (2) intrinsic affinity of \( \text{bbTBA} \) for the open and closed states is equal (which our data suggest is not entirely accurate; see Figs. 8–10).

Solutions
The pipette (external) solution contained (in mM) 140 KMeSO₃, 20 HEPES, 2 KCl, and 2 Mg²⁺ (pH 7.2). The bath (internal) solution contained (in mM) 136 KMeSO₃, 20 HEPES, 6 KCl, CaCl₂, and HEDTA. The amount of Ca²⁺ and HEDTA added to the internal solution was varied by estimating the amount of CaCl₂ and buffer required to yield a final Ca²⁺ concentration with MAXCHELATOR software (Bers, 1982). Thus, solutions of 0–4 \( \mu \)M free Ca²⁺ were buffered with 2 mM HEDTA, and those containing 10–20 \( \mu \)M free Ca²⁺ contained 1 mM HEDTA, while 100 \( \mu \)M free Ca²⁺ solutions were not buffered. To minimize voltage-dependent block by contaminant barium, (+) −18-crown-6-tetracarboxylic acid (18GTA) was added to the internal solution just before recording at a final concentration of 40 \( \mu \)M (Cox et al., 1997). Both 18GTA and \( \text{bbTBA} \) were prepared as millimolar stocks in internal solution and stored at −20°C before diluting into fresh solutions immediately before beginning experiments. Each patch was constantly bathed with internal solution using a sewer pipe system (DAD-12; Adams and List Assoc., Ltd.), which achieves solution changes in <1 s.

\( \text{N}-(4-[\text{benzoyl}]\text{benzyl})-\text{N},\text{N},\text{N}’\text{tributylammonium} \) bromide (\( \text{bbTBA} \)) was acquired from Spectra Group Limited, Inc. All experiments were performed using dilutions of an initial 50 mM stock prepared from newly obtained salt, which was stored frozen in small aliquots at −20°C. We did notice a slow decline in efficacy of block over a period of ~9 mo. This apparent degradation of \( \text{bbTBA} \) was independent of storage in solution (50 mM stocks frozen at −20°C) or as a salt, based on comparison of \( K_b \) at +100 mV of (a) the original stock solution, 3.2 \( \mu \)M; (b) the same stock solution 9 mo later, 4.4 \( \mu \)M; and (c) a freshly made solution prepared from 9-mo-old salt, 4.4 \( \mu \)M. The latter measurements (b and c) were made from Hill fits of dose–response data obtained from single patches, \( n = 3 \) patches. The absorption spectra (measured after 9 mo) for the latter two solutions were also comparable (\( A_{\text{max}} = 0.20 \) or 0.18, respectively); no measurement had been made for the original stock solution 9 mo prior. This slow decline in efficacy of block adds some variability to our averages, but will otherwise not affect the conclusions of this work, as the fractional block experiments were all performed in an ~3-mo period, and all comparisons were made to control experiments performed concurrently. Tetrabutylammonium (TBA) was purchased from Alfa Aesar. All other chemicals were purchased from Sigma-Aldrich.

RESULTS
Macroscopic \( \text{bbTBA} \) Blockade of BK Channels
Is Time Dependent
To characterize blockade of BK channels by \( \text{bbTBA} \), we first performed dose–response experiments, where macroscopic \( K^+ \) currents were elicited from inside-out macropatches using a series of voltage steps in the absence and presence of varying concentrations of \( \text{bbTBA} \) (Fig. 1). Fig. 1 A illustrates blockade of BK currents measured at +140 mV. \( \text{bbTBA} \) block was dose dependent and fully reversible (see Figs. 3 and 9). In Fig. 1 B, average fractional block, calculated from steady-state ionic currents in the presence and absence of \( \text{bbTBA} \), was plotted as a function of \( \text{bbTBA} \) concentration for
each of six test voltages in order to determine the dosage for half-maximal block (K_d). Hill fits of these data yielded K_d values in the low micromolar range, with Hill coefficients near 1. The resulting K_d values were then plotted as a function of test voltage, for voltages ≥ +40 mV (Fig. 1 C), where open probability (P_o) in 100 μM Ca^{2+} is expected to saturate. As expected for a charged molecule (see structure in Fig. 1 B), block was modestly voltage dependent, with a K_d at 0 mV of 9.5 μM, and a total voltage dependence zδ of 0.15, similar to other QA blockers in BK channels (Li and Aldrich, 2004). The lesser degree of voltage dependence compared with QA block of Kv channels (zδ/H11601 ∼ 0.4) may reflect the larger diameter, and presumed lower resistance, of the vestibule in BK channels. Fig. 1 demonstrates that bbTBA behaves as a typical QA blocker, where dose-dependent blockade results from binding of bbTBA to a single voltage-dependent binding site.

A fast time-dependent component of blockade was visible under certain recording conditions where activation is sufficiently fast to allow resolution of block kinetics (see Fig. 1 A). This component of block is examined in more detail in Fig. 2. Representative current families elicited by test potentials ranging from +20 to +120 mV in the absence (Fig. 2 A) and presence (Fig. 2 B) of 5 μM bbTBA illustrate the voltage range over which the time-dependent component is visible in 100 μM Ca^{2+} internal solution. Because BK channel activation is slowed in the absence of Ca^{2+} (Cox et al., 1997), the lowest concentration of Ca^{2+} for which time-dependent block was reliably observed (at extreme positive test potentials) was ~4 μM Ca^{2+} (see Fig. 7). Fig. 2 C shows, on a faster time scale, time-dependent block at bbTBA concentrations ranging from 1 to 20 μM at +180 mV in 100 μM Ca^{2+} solutions. Note that the time constant of the current relaxation is well fitted by a single exponential function (dashed lines) and is linearly related to bbTBA concentration in this range (Pearson’s R = 0.998; unpublished data), consistent with a bimolecular blocking reaction.

Time-dependent blockade by QA ions often signifies open channel block (Scheme 2); however, it is important to note that the implication is not limited to this possibility.

\[ C \rightleftharpoons O \rightleftharpoons B \]  
(Scheme 2)

The approximately twofold change in fractional block (estimated from the peak and steady-state levels of current in bbTBA) during the time-dependent component is well within the range that could be explained by voltage-dependent, state-independent block (Scheme 3) alone (assuming zδ ∼ 0.15, which predicts an approximately threefold change in block for the 260-mV step shown in Fig. 2 C), without a need to invoke state-dependent block.

![Figure 2](https://example.com/figure2.png)

**Figure 2.** bbTBA blockade is time dependent. (A) Representative current families (averages of three consecutive sweeps) measured in 100 μM internal Ca^{2+} in the absence and (B) presence of 5 μM bbTBA. Test voltages ranged from +20 to +120 mV (in increments of 20 mV). Holding potential was −80 mV. In saturating Ca^{2+}, the time-dependent component of block first becomes obvious at ~+80 mV. (C) Average currents at +180 mV in the absence and presence of the indicated concentrations of bbTBA, on an expanded time scale. The time-dependent component of blockade is visible at bbTBA concentrations between ~1 and 100 μM. The time constants of current relaxation decreases in rough proportion to the concentration of blocker applied, consistent with expectations of a two-state blocking reaction.

The bbTBA Binding Site Is Located in the Pore
Before using bbTBA to evaluate accessibility of the pore region of the BK channel, it is important to confirm
that the bbTBA binding site is located at the pore. We expect that bbTBA is a pore blocker since block is voltage dependent and bbTBA is structurally analogous to organic cations such as TBA, for which evidence of pore block is strong. To provide direct evidence of pore block, we evaluated the ability of the known pore blocker TBA to compete with bbTBA for binding to the BK channel. The experiment was conducted as depicted in Fig. 3 A; ionic currents at +120 mV were measured from channels before, during, and after exposure to 5 μM bbTBA, 1 mM TBA, or a solution containing both blockers (5 μM bbTBA + 1 mM TBA). Traces averaged from five consecutive steps (Fig. 3 B) were then used to calculate the percent block observed in each blocker solution. The average percent block from three patches is shown in Fig. 3 C. If both blockers can bind and block channels simultaneously (noncompetitive block; Scheme 4), we expect the degree of block by bbTBA + TBA to be 80%; however, we observe that block is weaker (71%), exactly what one can predict from a model of competitive block (Scheme 5) where bbTBA and TBA compete for a single binding site.

The data in Fig. 3 B were compared with the predictions of our final model (described in Materials and methods and Table I) to assess its adequacy in describing steady-state and kinetic properties of bbTBA blockade, and to illustrate the predictions made in Fig. 3 C. Blockade of ionic current by bbTBA was well described by our final model, as were currents in TBA (using estimates of block kinetics from Li and Aldrich [2004]; solid lines). The degree of blockade observed in the presence of both blockers matched predictions of the competitive model (Scheme 5, solid line), but was not compatible with non-competitive block (Scheme 4, dashed line). Note that the virtual absence of time dependence can be accounted for by both models. These data, along with the bimolecular kinetics and voltage dependence, provide good evidence that bbTBA is a pore blocker, as expected.
Figure 4. Kinetic and steady-state measurements of microscopic block at +100 mV are consistent with a bimolecular blocking reaction. 

(A) Representative single channel recordings (200 ms) measured at +100 mV in 100 μM internal Ca^{2+} in the absence or presence of the indicated concentration of bbTBA. The effective filtering frequency used for display purposes only was 1.96 kHz (for analysis of dwell times, records were sampled at 100 kHz and filtered at 10 kHz). Openings are upward. (B) Shut (left) or open (right) dwell time histograms were prepared from idealized records of control (top), 1 μM bbTBA (middle), or 3 μM bbTBA (bottom) data. Records were idealized with half amplitude threshold analysis using TAC software. Each histogram was constructed from 30–70 s of data, containing 1722–5081 events. Dwell times shorter than twice the filter dead time 2T_d (76 μs) were excluded. Histogram data (gray) plotted as square root of counts vs. log of dwell times were fitted with single exponential functions (solid lines) in order to determine mean dwell times for each condition. (C) Average rates of block at +100 mV were calculated from the dwell time histograms and plotted as a function of bbTBA concentration. The off rate (k_{off}, triangles) at each concentration of bbTBA was taken to be the reciprocal of the mean shut dwell time in blocker, assuming that excursions to the open level in the record largely reflect unblocking events. The first order on rate (k_{on}, squares) was calculated at each concentration by subtracting the native closing rate, 1/β (the reciprocal of the control mean open time), from the reciprocal of the mean open time in bbTBA to prevent contamination of the calculated on rate by the intrinsic rate of channel closing. Data from three patches at +100 mV were averaged (±SEM) and fitted with linear equations to yield the second order association rate (±95% confidence interval) k_{on} = 1.16 \times 10^{8} ± 0.258 \times 10^{8} M^{-1}s^{-1}. The dissociation rate (average of all values ± SEM) was k_{off} = 435 ± 28 s^{-1}, and the K_d (the intersection where k_{on} = k_{off}) was K_d = 3.9 μM. (D) Microscopic K_d was also obtained from steady-state measurements of open probability in various blocker concentrations. Open probabilities calculated from raw amplitude histograms were normalized, averaged, and plotted as a function of bbTBA concentration (n = 4 patches, ±SEM). The average data were fitted with the Hill equation, which yielded parameters (±95% confidence interval) K_d at +100 mV = 2.9 ± 5.5 μM and Hill coefficient of 1.0 ± 2.5. The linear increase in k_{on} as a function of bbTBA concentration and the independence of k_{off} are consistent with a bimolecular blocking reaction where a single bbTBA molecule binds at a single site.

Microscopic Blockade
To evaluate more carefully the kinetics and voltage dependence of bbTBA blockade, we performed microscopic experiments (Fig. 4). Fig. 4 A illustrates representative single channel records from inside-out patches at +100 mV in 100 μM Ca^{2+} internal solution, in the absence and presence of either 1 or 3 μM bbTBA. While the BK channel activity is characterized by high open probability (P_o) and short flickery closings, P_o decreases and open times become progressively shorter as the concentration of bbTBA is increased. Dwell time histograms constructed from idealized records were fitted with single exponential functions in order to yield the mean open and shut dwell times (Fig. 4 B). Block rates calculated from these mean dwell times were then plotted as a function of bbTBA concentration in order to yield the second order k_{on} at +100 mV, the average k_{off}, and the resultant K_d (Fig. 4 C). As an alternative to the microscopic measurements, we also determined the microscopic K_d using steady-state measurements of open probability in various bbTBA concentrations (Fig. 4 D). The results from Hill fits to normalized P_o plotted as square root of counts vs. log of dwell times were fitted with single exponential functions (solid lines) in order to determine mean dwell times for each condition. (C) Average rates of block at +100 mV were calculated from the dwell time histograms and plotted as a function of bbTBA concentration. The off rate (k_{off}, triangles) at each concentration of bbTBA was taken to be the reciprocal of the mean shut dwell time in blocker, assuming that excursions to the open level in the record largely reflect unblocking events. The first order on rate (k_{on}, squares) was calculated at each concentration by subtracting the native closing rate, 1/β (the reciprocal of the control mean open time), from the reciprocal of the mean open time in bbTBA to prevent contamination of the calculated on rate by the intrinsic rate of channel closing. Data from three patches at +100 mV were averaged (±SEM) and fitted with linear equations to yield the second order association rate (±95% confidence interval) k_{on} = 1.16 \times 10^{8} ± 0.258 \times 10^{8} M^{-1}s^{-1}. The dissociation rate (average of all values ± SEM) was k_{off} = 435 ± 28 s^{-1}, and the K_d (the intersection where k_{on} = k_{off}) was K_d = 3.9 μM. (D) Microscopic K_d was also obtained from steady-state measurements of open probability in various blocker concentrations. Open probabilities calculated from raw amplitude histograms were normalized, averaged, and plotted as a function of bbTBA concentration (n = 4 patches, ±SEM). The average data were fitted with the Hill equation, which yielded parameters (±95% confidence interval) K_d at +100 mV = 2.9 ± 5.5 μM and Hill coefficient of 1.0 ± 2.5. The linear increase in k_{on} as a function of bbTBA concentration and the independence of k_{off} are consistent with a bimolecular blocking reaction where a single bbTBA molecule binds at a single site.
bbTBA Does Not Behave as a Classical Open Channel Blocker for BK Channels

If BK channels posses an intracellular gate that occludes the bbTBA binding site when closed, one might predict that blocker binding and gate closing are mutually exclusive events, and that bound blocker would prevent or hinder closing of the activation gate. This is exactly what Armstrong originally observed for squid K⁺ channels (Armstrong, 1971), and is commonly observed for state-dependent blockers. In Fig. 6, we ask whether bbTBA alters the rate of channel closing for BK channels.

Tail currents measured from channels in the presence of 5 μM bbTBA were well fitted by single exponential functions (Fig. 6 A), although fits did deviate slightly from single exponential kinetics at potentials positive to −80 mV (discussed below). In contrast to expectations for state-dependent block, where bbTBA is predicted to slow the macroscopic rate of deactivation by ~2.3-fold, Fig. 6 B illustrates that bbTBA had no obvious effect on deactivation time constants over a wide range of potentials, although a progressive, mild slowing is apparent at positive potentials above −120 mV (1.1–1.4-fold). Although initially counterintuitive, our final state-independent model does in fact predict a progressive 1.2–1.3-fold slowing of deactivation at potentials between −200 and −50 mV, caused by the tendency for some open channels (either blocked or unblocked) to either unblock (OB-O-C) or block (O-OB-CB) before closing, rather than close directly in a single step (OB-CB or O-C). Thus, the kinetics of bbTBA block are such that although unblock does not require opening (in this model), a significant fraction of channels tend to traverse the same two-step closing pathways that state-dependent block would require, especially at positive potentials. TBA is known to speed deactivation (Li and Aldrich, 2004), an effect that was recapitulated in a single patch shown in Fig. 6 C, illustrating that the approximately twofold speeding of deactivation by TBA reported previously was
easily distinguishable in our measurements. The results of Fig. 6 suggest that closure of the activation gate is not physically hindered by the presence of bbTBA, which is consistent with at least two alternative interpretations; either closure of the gate does not occlude the bbTBA binding site (state-independent block) or the bbTBA binding site is gated, but channels close with a blocker molecule trapped inside the pore (an alternative model of state-dependent blockade).

Some channels are capable of trapping small molecules inside the pore cavity after closure of the activation gate (Armstrong, 1971; Strichartz, 1973; Neely and Lingle, 1986; Holmgren, Smith and Yellen, 1997). This would be represented by extension of the state-dependent model (Scheme 2) into C-O-OB-CB. Modeling confirms that this trapping model could also explain the mild slowing of deactivation by bbTBA and still assume that block is state dependent. Indeed, there is evidence that the internal cavity of BK channels is larger than that of most K\(^{+}\) channels (Li and Aldrich, 2004), and thus may be large enough to accommodate bbTBA with a closed (cytoplasmic) gate. If BK channels can trap bbTBA, then one would expect that a substantial fraction of open channels that are blocked during the test pulse will remain blocked immediately after rapid closure and reopening, and current will simply rise to steady-state block levels with minimal or no time dependence. However, we found no evidence for trapping of bbTBA inside BK channels (Fig. 7). Fig. 7 A shows a trapping experiment performed in 20 \(\mu\)M internal Ca\(^{2+}\), where three control pulses to +150 mV were followed by three pulses in the presence of 5 \(\mu\)M bbTBA. The interpulse interval was set to 5 ms at −200 mV, in order to close channels rapidly and minimize the possibility of bbTBA escaping from open channels, either before closing or during rare reopening events (\(P_o\) at −200 mV ∼ 0.0001; Horrigan and Aldrich, 2002). Fig. 7 A shows that time-dependent block remained intact for each test pulse. In the context of a state-dependent model of block, the persistence of the time dependence would suggest that channels do not remain blocked during the interpulse interval, i.e., channels cannot trap bbTBA and have to wait for blocker to exit before closing (in which case one would expect to see slowing of deactivation). Obviously, this interpretation is difficult to reconcile with Fig. 6. Alternatively, a state-independent model of block posits that time dependence simply reflects equilibration of channels to the change in test potential; one would not expect this to change for three identical test pulses.

The trapping experiment was repeated in 4 \(\mu\)M internal Ca\(^{2+}\) (Fig. 7 B) where probability of channel opening at −200 mV is expected to be 10-fold lower than in 20 \(\mu\)M Ca\(^{2+}\) (∼ 0.00001; Horrigan and Aldrich, 2002). The apparent inability of channels to trap bbTBA (and the lack of slowed deactivation at negative potentials in Fig. 6) cannot be explained by rapid escape of bbTBA before channels close, since the expected rate of unblock at −200 mV (1234 s\(^{-1}\); extrapolated from Fig. 5 B) is roughly four times slower than the rate of channel

![Figure 6](http://example.com/fig6.png)  
**Figure 6.** bbTBA does not interfere with closure of the activation gate. (A) An instantaneous IV protocol was used to measure rates of channel deactivation in the absence and presence of 5 \(\mu\)M bbTBA. Channels were maximally activated by a step from −80 to +120 mV, and then returned to a series of repolarization potentials ranging from −200 to −30 mV. Representative tail currents elicited by repolarization to −100 mV in the absence (black) and presence (gray) of 5 \(\mu\)M bbTBA are illustrated. Tail currents were fitted with single exponential functions (dashed lines) in order to obtain deactivation time constants. (B) Deactivation time constants from 14 patches in the absence (black) or presence (gray) of 5 \(\mu\)M bbTBA were averaged (±SEM) and plotted as a function of repolarization voltage. (C) Deactivation time constants from a single patch comparing control (circles) with 5 \(\mu\)M bbTBA (triangles), 1 mM TBA (squares), or both blockers together (bowties) illustrate the speeding effect of TBA on deactivation, which persists in the presence of both blockers. The mild slowing of deactivation kinetics at positive potentials can be explained by the tendency of channels at these potentials to close via the same two-step pathway that state-dependent models would require.
Figure 7. BK channels show no evidence of trapping bbTBA inside the cavity after closure. (A) To test for trapping, channels were stepped from a holding potential of −200 mV to a test potential of +150 mV three times, separated by a 5-ms interpulse interval at −200 mV, first in a control solution containing 20 μM internal Ca^{2+} (black traces), and then in 5 μM bbTBA (gray traces). Traces were then superimposed (one control sweep, average of three sweeps in blocker) for comparison. (B) The trapping experiment was repeated in 4 μM internal Ca^{2+}, where open probability at −200 mV was roughly an order of magnitude lower (<0.00001; Horrigan and Aldrich, 2002), in order to minimize the possibility of escape of blocker from channels reopening during the 5-ms repolarization to −200 mV. In this experiment, the test potential was +200 mV, and the traces shown are averages of three (control) or four (bbTBA) sweeps. This experiment in 4 μM Ca^{2+} was repeated in eight different patches. (C) Fractional unblock (1 − I_{bbTBA}/I_{control}) was measured in 100 μM internal Ca^{2+} either from steady-state currents (black triangles; n = 19 patches ± SEM) or peak tail currents (gray squares; n = 4 patches ± SEM) and plotted as a function of test voltage. The steady-state data are the high Po measurements displayed in Fig. 8 C. Steady-state data were fit with Eq. 3 either by Igor (dashed line; parameters ± 95% confidence interval were K_{0}(0) = 8.3 ± 1.2 μM and δo = 0.14 ± 0.03) or by our final model (thick black line, K_{0}(0) = 8.5 μM and δo = 0.15). The final model fit was then extrapolated to negative potentials. Fractional unblock computed from tail unblock was also fitted (thin gray line) with parameters (±95% confidence interval): K_{0}(0) = 2.9 ± 1.8 μM and δo = 0.12 ± 0.016. The observed level of block at negative potentials is much greater than steady-state predictions, indicating that bbTBA has not dissociated from channels at the time of closing. The persistence of time-dependent block after closure and reopening of blocked channels suggests that bbTBA is able to dissociate from closed channels during the 5-ms repolarization.

bbTBA Blockade Does Not Depend Strongly upon Open Probability

One can directly assay the state dependence of channel blockade by comparing the efficacy of block under conditions where P_o is high and low, while voltage is kept constant. The BK channel is particularly amenable to such experiments by virtue of its activation by both voltage and Ca^{2+}. Thus, one can measure ionic current (and blockade) at a single test potential while activating channels to different levels of open probability with Ca^{2+}. If bbTBA block requires channel opening, steady-state block should be maximal at high P_o and minimal at low P_o where only a few channels have reached the open state. Conversely, block that is independent of channel conformation should be equally effective at all levels of P_o. To discriminate between these possibilities, macroscopic steady-state fractional block was measured at several test potentials under a variety of different Ca^{2+} conditions representing a wide range of open probabilities (Fig. 8). Fig. 8 A illustrates two representative measurements of ionic current blockade at +90 mV from the same patch exposed to either 100 or 1 μM internal Ca^{2+}, ±5 μM bbTBA. The relative P_o vs. voltage relationships for this patch (Fig. 8 B) show that open probability at this voltage is ~1 or ~0.25 under these conditions, respectively. It is apparent in Fig. 8 A that despite the large difference in open probability, fractional block appears roughly equivalent (although block...
tance was calculated from tail current amplitudes measured at −80 mV, and then normalized and plotted as a function of test potential. Resulting P_o-V curves were fit with a Boltzmann equation (Eq. 4) to yield parameters (±95% confidence interval): 100 μM Ca^{2+} (control/ (+bbTBA), V_1/2 = −40 ± 1.5 mV/−42 ± 2.0 mV and z = 1.7 ± 0.15/2.1 ± 0.32; 1 μM Ca^{2+} (control/ (+bbTBA), V_1/2 = −113 ± 1.2 mV/+107 ± 2.6 mV, z = 1.5 ± 0.9/1.4 ± 0.18. Fractional unblock (I_{bbTBA}/I_{control}) was computed from steady-state currents of ±5 μM bbTBA exposed to 1 μM internal Ca^{2+} (circles) or 100 μM internal Ca^{2+} (triangles), plotted as a function of voltage and fitted with Eq. 3, yielding the parameters (±95% confidence interval): 100 μM Ca^{2+}, K_0(0) = 8.9 ± 0.6 μM and zδ = 0.13 ± 0.03; 1 μM Ca^{2+}, K_0(0) = 9.1 ± 3.6 μM and zδ = 0.09 ± 0.08. Measurements in A and B were all made in the same patch. (C) Relative P_o and fractional unblock (I_{bbTBA}/I_{control}) were measured over a range of test potentials from each of 24 patches in solutions ranging from 0 to 100 μM Ca^{2+}, and measurements of fractional unblock at each voltage were binned into P_o categories of “low” (0 < P_o < 0.2; crosses), “medium” (0.4 < P_o < 0.6; circles), and “high” (0.8 < P_o < 1.0; triangles). Within each P_o category, fractional unblock was averaged at each test potential, plotted ± SEM as a function of voltage, and fitted with Eq. 3 (dashed lines) to yield parameters (±95% confidence intervals): low P_o, K_0(0) = 10.1 ± 5.6 μM and zδ = 0.17 ± 0.13; medium P_o, K_0(0) = 8.1 ± 2.8 μM and zδ = 0.16 ± 0.09; high P_o, K_0(0) = 8.3 ± 1.2 μM and zδ = 0.14 ± 0.05. A fit to the final model of bbTBA block (K_0(0) = 8.9 μM and zδ = 0.15) is also illustrated with a solid line. (D) The same fractional unblock data were then binned into three separate test voltages: +60 mV (circles), +100 mV (triangles), and +150 mV (bowties), and each data point plotted individually versus relative P_o, estimated from the P_o-V curve for that patch. Data points within each voltage category were then fitted with a linear equation. Macroscopic fractional unblock does not depend strongly upon open probability, although block may be slightly less effective at low P_o.

Figure 8. Efficacy of macroscopic bbTBA blockade appears largely independent of open probability. (A) Representative BK currents measured from a single patch at +90 mV from a holding potential of −80 mV, ±5 μM bbTBA in either 100 μM internal Ca^{2+} (P_o ~ 1) or 1 μM internal Ca^{2+} (P_o ~ 0.25); fractional block computed from the steady-state currents was 50.1% or 44.7%, respectively. (B) Relative P_o was estimated from normalized conductance voltage relationships obtained in the absence (filled symbols) and presence (open symbols) of 5 μM bbTBA, for both 1 μM internal Ca^{2+} (circles) or 100 μM internal Ca^{2+} (triangles) solutions. Relative conductance was calculated from tail current amplitudes measured at −80 mV, and then normalized and plotted as a function of test potential. Resulting P_o-V curves were fit with a Boltzmann equation (Eq. 4) to yield parameters (±95% confidence interval): 100 μM Ca^{2+} (control/ (+bbTBA), V_1/2 = −40 ± 1.5 mV/−42 ± 2.0 mV and z = 1.7 ± 0.15/2.1 ± 0.32; 1 μM Ca^{2+} (control/ (+bbTBA), V_1/2 = −113 ± 1.2 mV/+107 ± 2.6 mV, z = 1.5 ± 0.9/1.4 ± 0.18. Fractional unblock (I_{bbTBA}/I_{control}) was computed from steady-state currents of ±5 μM bbTBA exposed to 1 μM internal Ca^{2+} (circles) or 100 μM internal Ca^{2+} (triangles), plotted as a function of voltage and fitted with Eq. 3, yielding the parameters (±95% confidence interval): 100 μM Ca^{2+}, K_0(0) = 8.9 ± 0.6 μM and zδ = 0.13 ± 0.03; 1 μM Ca^{2+}, K_0(0) = 9.1 ± 3.6 μM and zδ = 0.09 ± 0.08. Measurements in A and B were all made in the same patch. (C) Relative P_o and fractional unblock (I_{bbTBA}/I_{control}) were measured over a range of test potentials from each of 24 patches in solutions ranging from 0 to 100 μM Ca^{2+}, and measurements of fractional unblock at each voltage were binned into P_o categories of “low” (0 < P_o < 0.2; crosses), “medium” (0.4 < P_o < 0.6; circles), and “high” (0.8 < P_o < 1.0; triangles). Within each P_o category, fractional unblock was averaged at each test potential, plotted ± SEM as a function of voltage, and fitted with Eq. 3 (dashed lines) to yield parameters (±95% confidence intervals): low P_o, K_0(0) = 10.1 ± 5.6 μM and zδ = 0.17 ± 0.13; medium P_o, K_0(0) = 8.1 ± 2.8 μM and zδ = 0.16 ± 0.09; high P_o, K_0(0) = 8.3 ± 1.2 μM and zδ = 0.14 ± 0.05. A fit to the final model of bbTBA block (K_0(0) = 8.9 μM and zδ = 0.15) is also illustrated with a solid line. (D) The same fractional unblock data were then binned into three separate test voltages: +60 mV (circles), +100 mV (triangles), and +150 mV (bowties), and each data point plotted individually versus relative P_o, estimated from the P_o-V curve for that patch. Data points within each voltage category were then fitted with a linear equation. Macroscopic fractional unblock does not depend strongly upon open probability, although block may be slightly less effective at low P_o.
Concentration was calculated as $1 - \frac{P_{o\text{bbTBA}}}{P_{o\text{control}}}$. Fractional block for a few experiments was also computed from bursts of activity ± bbTBA, such as those illustrated in B and D on an expanded time scale; percent block from these analyses was within 5% of that calculated from the complete dataset. Long closures were occasionally omitted from records. (B) Representative sweeps illustrating block at low $P_o$ (sweeps 1 and 2 in A). The top row illustrates 4 s of recording in 2 μM internal Ca$^{2+}$ (sweep 1) and 2 μM Ca$^{2+}$ + 5 μM bbTBA (sweep 2). Average $P_o$ under these conditions for this experiment was 0.09 or 0.05, respectively. 100 ms of this record shown on an expanded time scale below each trace on the same expanded time scale as in B in order to facilitate comparison with the traces in B. (C) Fractional block was calculated from the average raw amplitude histograms for the control (solid line) and 5 μM bbTBA (dashed line) data. For this experiment, at $P_o = 0.09$, fractional block was 44.5%. (D) Block at high $P_o$ is illustrated by 500 ms of recording in 100 μM internal Ca$^{2+}$ (sweep 3) or 100 μM Ca$^{2+}$ + 5 μM bbTBA (sweep 4), where $P_o$ was 0.89 or 0.44, respectively. 100 ms of this record (indicated by brackets) are shown below each trace on the same expanded time scale as in B in order to facilitate comparison with the traces in 2 μM Ca$^{2+}$. (E) Fractional block computed from the raw amplitude histograms for this experiment at $P_o = 0.9$ was 51.2%. The efficacy of microscopic block appears to decrease by ~6% at low $P_o$. It is shown in Fig. 9 A. 10 s sweeps at +40 mV were recorded from an excised patch perfused with 2 μM internal Ca$^{2+}$, which was alternated with 5 μM bbTBA several times before repeating the process with 4, 10, and 100 μM Ca$^{2+}$ solutions ± 5 μM bbTBA. This procedure allowed us to monitor directly the stability of $P_o$ over the duration of the experiment and ensure accurate measurement of $P_o$ changes induced by blocker. The first two sweeps in the absence and presence of bbTBA for 2 μM Ca$^{2+}$ (sweeps 1 and 2 in Fig. 9 A) are illustrated in Fig. 9 B. As shown in the top left panel of Fig. 9 B, $P_o$ in 2 μM Ca$^{2+}$ was 0.09; 100 ms of this record shown on an expanded time scale below illustrates that channel activity mainly consisted of short openings punctuated by flickery closings. When 5 μM bbTBA was applied, $P_o$ was reduced to 0.05, primarily due to a reduction in mean open time (Fig. 9 B, right). Fractional block at $P_o \sim 0.09$ computed from the raw amplitude histograms in Fig. 9 C was 45%. Sweeps 3 and 4 recorded in 100 μM internal Ca$^{2+}$ ± 5 μM bbTBA are illustrated in Fig. 9 D. In this case, control $P_o$ (0.89) was reduced to 0.44 in the presence of bbTBA. Note from the data on a faster time scale that channel activity is qualitatively similar to that observed in 2 μM Ca$^{2+}$; control openings are still interrupted by flickery closings, and mean open time in blocker is
similarly reduced. Fractional block computed from the amplitude histograms in Fig. 9 E at $P_o \sim 0.9$ was 51%. This experiment shows that, as the macroscopic data suggested, block does appear modestly reduced at low $P_o$. To see whether this $\sim 6\%$ decrease in block at low $P_o$ is compatible with state-dependent block, we compared data from several patches to model predictions in Fig. 10.

Individual measurements of microscopic fractional block from several experiments were plotted as a function of $P_o$ (Fig. 10). Predictions of fractional block made using state-dependent (Scheme 2) and state-independent (Scheme 3) models are shown as dashed lines. The data are best fit by a line that matches closely the predictions of purely state-independent block, it is clearly not compatible with the large change in apparent affinity predicted by the state-dependent model for the same change in $P_o$ (which predicts a 31-fold lowering of apparent affinity, giving a $K_a$ of 144 $\mu$M at $P_o = 0.03$).

The mechanistic picture of bbTBA blockade that emerges from Figs. 8–10 is where binding of bbTBA is enhanced to a small degree in the open state, but not gated in an absolute fashion by channel conformation. The difference in apparent affinity between the open and closed conformations results in a modest 7% change in the fraction of blocked channels over a range of $P_o$, encompassing nearly two orders of magnitude. The final piece of evidence that we present against open channel block comes from returning again to the initial observation that bbTBA blockade is time dependent.

At high concentrations of bbTBA ($\sim 75–100$ $\mu$M), a third component of time-dependent block becomes apparent in the macroscopic record. Fig. 11 A illustrates the three kinetic phases of time-dependent block for currents at $+90$ or $+150$ mV in the presence of 75 $\mu$M bbTBA; current initially increases to a maximum, and then rapidly declines, and finally increases again to steady state. These complex kinetics are predicted by our final model of state-independent block, and, importantly, are not compatible with state-dependent block (Fig. 11 B). The data at $+90$ mV were roughly approximated by our final model (dashed black line). Only minor adjustment of those parameters was required to fit the steady-state level of blockade in this patch; we also added the assumption that bbTBA slows activation in order to more accurately describe the kinetics (solid red line). (It is important to note that, while we have no evidence to justify this particular adjustment, this slowing of activation was not required in order to reproduce the triphasic kinetics of block.) The data require that some fraction of channels start in the closed, blocked state, consistent with expectations of voltage-dependent, state-independent block. The smaller fraction of closed, blocked channels required to fit our data (35–50%), compared with predictions of $\sim 80\%$ may be due to the preference of bbTBA for the open state. Finally, the state-dependent version of our final adjusted model is grossly incompatible with the observed data, predicting a much larger change in fractional block during the time-dependent component, and biphasic kinetics (solid blue line).

Fig. 11 C uses the occupancy of each state over time to explain how these complex kinetics arise in the ionic currents, and why they cannot be observed for state-dependent block. The final adjusted model is illustrated in the inset, including the rate constants used for modeling block at $+90$ mV in 75 $\mu$M bbTBA. Each state in the model is represented by a different color, and the fraction of channels populating that state is plotted over time in the same color. As in Fig. 11 B, we assume that 35% of channels are initially closed and blocked (CB), while the remaining 65% are closed (C). At the start of the voltage step, occupancy of the closed state falls as...
360 State-independent Block of BK Channels

many channels quickly become blocked (C-CB), and the remaining channels more slowly enter the open state (C-O). Thus, occupancy of CB rises and then begins to decline slowly as channels continue on to the open blocked state (CB-OB); this in turn is reflected by the steady rise in occupancy of OB. Meanwhile, those channels that entered the open state initially rapidly become blocked and contribute to the population of OB. Finally, the slow transition of channels from C to O comes to a halt, and occupancy reaches a steady state. Thus, occupancy of the open state (which shapes the kinetics of current blockade) is dominated in the first phase by channels entering the open state (C-O); during the second phase, channels leave the open state to become blocked; and finally in the third phase, the slow kinetics of the closed to open transition again dominate as occupancy of O and OB reaches steady-state equilibrium.

Triphasic block kinetics arise from the simultaneous combination of the relatively fast blocking process and the comparably slower activation process. The rapid population of the blocked states requires direct connectivity between the closed and blocked states (C and CB). This is in direct contrast with state-dependent models, where channel opening is the rate-limiting step for blockage, and explains why triphasic block kinetics cannot be obtained from state-dependent models. Although triphasic kinetics do not require some fraction of channels starting in CB, the correct amplitude of the peak current does. No permutation of state-dependent models, including a trapping model, was able to account for this complex time dependence. Fig. 11 provides strong support for a state-independent mechanism of bbTBA blockade of BK channels.

**DISCUSSION**

Time-dependent blockade, although often suggestive of open channel block, is equally compatible with models that do not require the open channel conformation for blocker to access its binding site. We found that the time dependence of bbTBA block is a direct result of rapid population of blocked states prior to equilibration of the open state, in direct contrast to state-dependent models that require that the slow C to O transition be rate limiting for block. The triphasic kinetics of block observed at high concentrations of bbTBA strongly suggest that blockade is not state dependent.
which predicts slowing at all potentials, especially negative voltages, and suggests that bound bbTBA does not prevent channel closing. (b) Trapping experiments suggest that bbTBA cannot be confined within the cavity after channel closure, indicating that closure of the activation gate does not prohibit bbTBA dissociation. (c) Macroscopic and microscopic measurements of steady-state block as a function of open probability suggest that block does in fact have a mild preference for the open channel conformation, but to a degree that deviates substantially from predictions of a purely state-dependent model. The observation that ~45% of channels are blocked when fewer than 10% of channels are open further supports the idea that channels can be blocked in the closed state. (d) Finally, our final state-independent model of bbTBA block was able to accurately describe the kinetics of blockade at high concentrations of bbTBA, where the time dependence of block was triphasic. These unusual triphasic kinetics, which result from simultaneous blocking and opening of channels, could not be reproduced by any state-dependent model we tested. Together, our results strongly suggest that closure of the activation gate does not present a significant barrier to bbTBA binding site access.

Implications for Location of the BK Channel Permeation Gate

Relatively unrestricted access of intracellular bbTBA in the open and closed channel conformations implies that K\(^{+}\) flux is gated at a location that is extracellular to the bbTBA binding site. Since QA ions are known to dock at the internal entrance to the selectivity filter, deep within the channel’s intracellular cavity (Choi et al., 1993; Zhou et al., 2001a; Lenaeus et al., 2005), this places the permeation gate at or above the selectivity filter. Our findings suggest that current assumptions about BK channel architecture and gating may need to be reevaluated.

Jiang et al. (2001, 2002a,b) postulate that mammalian BK channels share the same gating mechanism proposed for the bacterial Ca\(^{2+}\)-gated K\(^{+}\) channel MthK. Extrapolation of their gating model from the prokaryotic MthK to the eukaryotic BK channel was based on <20% amino acid conservation between the two proteins, and a shift in the voltage dependence of activation after mutation of a putative salt bridge site in the BK channel (Jiang et al., 2001). Their gating model involves an intracellular gating ring which, upon ligand binding, tugs the S6 bundle helices apart. Niu et al. (2004) have expanded upon this gating model by proposing that both Ca\(^{2+}\) and voltage activation of BK channels involves passive coupling of either the ligand-binding gating ring or the voltage-sensing apparatus to the S6 gate via separate hinges, based on the observation that deletions and insertions into the lower S6 region alter gating. However, mutations in the S6 regions of K\(_{ATP}\) and CNG channels, which do not appear to use intracellular S6 gates, also disrupt gating (Trapp et al., 1998; Flynn and Zagotta, 2003). In addition, Ca\(^{2+}\)-dependent gating of the MthK channel differs in significant ways from that of the BK channel, suggesting the possibility of different gating mechanisms (Zadek and Nimigean, 2006). Thus, in the absence of any direct evidence in support of an intracellular gate (or gating ring), the degree to which this model represents BK channel gating remains unclear.

Recent work (unpublished data) provides strong evidence that conformational rearrangements associated with gating regulate access of intracellular Shaker ball peptides to the BK channel pore. The support for this comes from time-dependent block, slowing of macroscopic deactivation and a hook in the tail currents, and correlation of the degree of fractional block with macroscopic \(P_o\). Although clearly demonstrating state-dependent block by ball peptides, that study is not conclusive about the gate for K\(^{+}\) ions, since these 20 amino acid peptides are orders of magnitude larger than permeant K\(^{+}\) ions. How can we reconcile gated access of a bulky peptide with relatively unrestricted access of a smaller quaternary ammonium derivative? One could speculate that permeation gating of BK channels involves movements at or above the selectivity filter that are obligatorily coupled to rearrangements of an intracellular protein structure (for example, the putative gating ring). Reorganization of intracellular regions, including the S6 helices, may indeed gate access of ball peptide to the pore, while for the smaller blocker bbTBA, these same movements only slightly alter access and/or binding to a constantly available site. The reverse conformational change in S6, while permitted by bound bbTBA, could be sterically prohibited until ball peptide dissociates, resulting in slower macroscopic deactivation.

Might BK Channels Have a Selectivity Filter Gate?

Gating of K\(^{+}\) flux at the selectivity filter of BK channels is an attractive possibility, when one considers the classic experiments of Miller et al. (1987), which showed not only that occupancy of BK channel selectivity filter by Ba\(^{2+}\) destabilizes the closed state (Miller, 1987; Miller et al., 1987), but also that, in the closed state, the selectivity filter is inaccessible to Ba\(^{2+}\) from both the internal and external face of the membrane, as if both ends of the selectivity filter pinch shut (Miller et al., 1987). This possibility is made all the more attractive by recent work illustrating that BK channel gating is unusually sensitive to selectivity filter occupancy; permeant thallium favors flickery gating on the submillisecond time scale, which may be a direct result of rapid conformational changes of the selectivity filter (Piskorowski and Aldrich, 2006), and also prevents TBA from speeding deactivation (Piskorowski, 2005).
One of the best studied selectivity filter gates is the C-type inactivation gate of Kv channels. Functional evidence strongly suggests that C-type inactivation involves conformational changes of the outer mouth and selectivity filter, as well as the S4 voltage-sensing region, of Kv channels (Timpe et al., 1988; Choi, Aldrich and Yellen, 1991; Lopez-Barneo et al., 1993; Yellen et al., 1994; Baukrowitz and Yellen, 1996; Loots and Isacoff, 1998). A similar mechanism may account for voltage-dependent gating of Kcsa (Cordero-Morales et al., 2006). A rather striking feature of BK channels, by contrast, is their lack of intrinsic inactivation, even after long periods of recording. A simple explanation for this might be that the C-type inactivation gate for BK channels is in fact the permeation gate itself.

Comparisons with CNG Channel Gating

Gating of CNG channels, which are also activated by intracellular ligand and are evolutionary relatives of BK channels, embodies many of the basic principles we propose for BK channels. In general, activation appears to involve a substantial rearrangement of the lower S6 bundle crossing, as evidenced by state-dependent block by intracellular tetracaine (Fodor et al., 1997a,b), dequalinium (Rosenbaum et al., 2004), and state-dependent QA block and accessibility (Contreras and Holmgren, 2006). These motions, which widen the constriction from ~5 to 12 Å (Flynn and Zagotta, 2001; Flynn and Zagotta, 2003), are required to accommodate both spontaneous salt bridge formation at lower S6 residues only in the closed state, and the large difference in reactivity of the relatively large MTSET molecule with pore-lining residues in the open and closed states (Flynn and Zagotta, 2001). However, constriction of the internal pore entrance does not constitute permeation gating for CNG channels: smaller Ag+ and MTSEA ions react roughly equivalently with pore residues regardless of channel conformation (Flynn and Zagotta, 2001), and pore block by Ni2+ is state independent (Karpen et al., 1993). Rather, it is thought that S6 conformational changes are coupled to intrinsic permeation gating of CNG channels at the selectivity filter, a proposal that draws support from state-dependent conformational changes observed in the outer pore region (Becchetti and Roncaglia, 2000; Liu and Siegelbaum, 2000).

CNG channels also exemplify that the apparent affinity of blockers can depend, often dramatically, on channel conformation, without being gated, e.g., enhanced coordination of Ni2+ by lower S6 histidine residues in the open state (Gordon and Zagotta, 1995a,b), and increased apparent affinity of several large blockers in the closed state (Fodor et al., 1997a,b; Rosenbaum et al., 2004). Similarly, block of BK channels by C10 showed a preference for the open state, which was far less than one would expect based on pure open channel block (Li and Aldrich, 2004). The observed slight preference of bbTBA for the open conformation in the present study appears similar.

Widening of the intracellular bundle crossing for both CNG and BK channels gates access of larger molecules, while smaller reagents and permeant ions are able to pass through the aperture regardless of channel state. If the largest molecules able to pass the bundle crossing freely are any indication of size, the diameter of the closed state bundle crossing is larger for BK channels than for CNG channels, consistent with the larger dimensions of the BK channel cavity (Li and Aldrich, 2004). We speculate that BK channels may share with their CNG cousins a conserved gating mechanism, which involves conformational changes at the bundle crossing and permeation gating at the selectivity filter.

Conclusions and Future Directions

Our work presents evidence for a voltage-gated channel that is not gated at the S6 bundle crossing. Although the exact location of the permeation gate for BK channels remains an open question, the most obvious candidate position is the selectivity filter itself. Despite numerous potential pitfalls (including 29 native cysteines, functional consequences of MTS modification, activation of BK channels by Cd2+, and buffering requirements for Ag+, which would stipulate minimal Ca2+ conditions), future experiments using cysteine accessibility and small probes (e.g., Ag+, Cd2+, or MTSEA) may be useful to convincingly establish the location of the permeation gate.

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REFERENCES


